

## A MINOR DITERPENE FROM *AMARACUS AKHDARENSIS*

FRANCO PIOZZI, MARIAPIA PATERNOSTRO and SALVATORE PASSANNANTI

Istituto di Chimica Organica, Università di Palermo, Archirafi 20, 90123 Palermo, Italy

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**Key Word Index**—*Amaracus akhdarensis*, Labiatae, diterpenoid, new isopimarane derivative, isoakhdatriol

**Abstract**—A new isopimarane diterpenoid, isoakhdatriol, was isolated in very small amount from the aerial part of *Amaracus akhdarensis*. Its structure, isopimar-15-en-3 $\beta$ ,8 $\beta$ ,19-triol, was established by spectroscopic means

### INTRODUCTION

We have recently reported [1] on the structure of three isopimarane diterpenoids isolated from *Amaracus akhdarensis* (Ietswaart et Boulos) Brullo et Furnari [2] (syn *Origanum akhdarensis*), Labiatae. The products were akhdarenol (isopimar-7,15-dien-19-ol), akhdardiol (isopimar-15-en-8 $\beta$ ,19-diol, 1) and akhdatriol (isopimar-15-en-8 $\beta$ ,11 $\alpha$ ,19-triol). From the acetone extract of this species a fourth product, isoakhdatriol, has been isolated now in very small amount: its structure is here elucidated as isopimar-15-en-3 $\beta$ ,8 $\beta$ ,19-triol (2).

### RESULTS AND DISCUSSION

Isoakhdatriol has formula  $C_{20}H_{34}O_3$ . The  $^1H$  NMR data indicate the occurrence of three *tert* methyls, one vinyl group, one primary and one secondary hydroxy function. The spectrum is very similar to that of akhdardiol, thus suggesting the same isopimarane skeleton, with three main differences. First, a proton at  $\delta$  3.42 (not present in 1) appears as a double doublet with  $J_{ax-ax}$  8 Hz and  $J_{ax-eq}$  4 Hz, indicating a C-CHOH-CH<sub>2</sub>- group with the hydroxy function in the equatorial position. Second, the separation of the two doublets of the CH<sub>2</sub>OH protons is greater than in akhdardiol and akhdatriol: indeed, the doublets are centred at  $\delta$  3.37 and  $\delta$  4.23, whereas they fall at  $\delta$  3.48 and  $\delta$  3.78 in 1 respectively.  $\delta$  3.55 and  $\delta$  3.90 in akhdatriol, such chemical shifts are observed [3] for 4 $\beta$ -CH<sub>2</sub>OH isopimaranes having a 3 $\beta$ -OH substituent. Third, the singlet attributed to 4 $\alpha$ -Me is shifted from  $\delta$  0.96 (akhdardiol) to  $\delta$  1.21 (isoakhdatriol). These data strongly suggest the occurrence of a 3 $\beta$ -OH group in isoakhdatriol.

The  $^{13}C$  NMR data of isoakhdatriol 2, compared with those of akhdardiol (1) (see Table 1), clearly confirm the occurrence of the secondary hydroxy group on C-3. Significant  $\beta$ -effects (downfield shifts) are observed for C-2 and C-4, while  $\gamma$ -effects (upfield shifts) occur for C-1 and C-18, and to a lesser extent for C-5 and C-19.

Hence isoakhdatriol has to be represented by the structure 2.

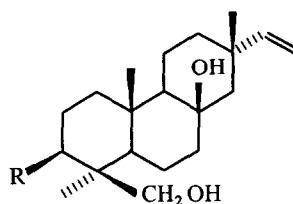
### EXPERIMENTAL

Plant material was collected by Prof. F. Furnari (Institute of Botany, University of Catania) in Northern Cyrenaica (Libya), on

the Gebel El Akhdar, in May 1981. Specimens are deposited in the Herbarium of the above Institute.

Extraction and chromatography of the diterpene fraction were described in a previous note [1]. Some fractions eluted with pet ether-AcOEt (60/40) and containing mainly akhdatriol revealed the co-occurrence of a product having close polarity. Semipreparative HPLC (Waters instrument,  $\mu$ -Porasil column 7.8 mm  $\times$  30 cm, eluant cyclohexane-EtOAc, 1:1, 2 ml/min) permitted the separation of akhdatriol (peak after 11'30") from isoakhdatriol (peak after 15'). The product which was isolated (6 mg, oily, homogeneous on HPLC) could not be crystallized.

High resolution MS found 322.257,  $C_{20}H_{34}O_3$  requires 322.250. MS  $m/z$  322, 307, 304, 289, 286, 273, 271, 223, 205.  $^1H$  NMR ( $CDCl_3$ , 300 MHz)  $\delta$  0.94 (s, 3H, 10 $\beta$ -Me), 1.21 (s, 3H,



- 1 R = H  
2 R = OH

Table 1  $^{13}C$  NMR chemical shifts of compounds 1 and 2

C	1	2	C	1	2
1	39.59 <sub>t</sub>	37.75 <sub>t</sub>	11	17.18 <sub>t</sub>	17.50 <sub>t</sub>
2	18.07 <sub>t</sub>	27.73 <sub>t</sub>	12	38.13 <sub>t</sub>	38.18 <sub>t</sub>
3	35.74 <sub>t</sub>	80.94 <sub>d</sub>	13	37.20 <sub>s</sub>	36.91 <sub>s</sub>
4	38.68 <sub>s</sub>	43.07 <sub>s</sub>	14	51.57 <sub>t</sub>	51.70 <sub>t</sub>
5	57.21 <sub>d</sub>	56.33 <sub>d</sub>	15	151.59 <sub>d</sub>	151.53 <sub>d</sub>
6	18.35 <sub>t</sub>	17.81 <sub>t</sub>	16	108.57 <sub>t</sub>	108.72 <sub>t</sub>
7	43.99 <sub>t</sub>	43.99 <sub>t</sub>	17	24.28 <sub>q</sub>	24.39 <sub>q</sub>
8	72.49 <sub>s</sub>	72.28 <sub>s</sub>	18	27.08 <sub>q</sub>	22.69 <sub>q</sub>
9	58.21 <sub>d</sub>	56.97 <sub>d</sub>	19	65.25 <sub>t</sub>	64.42 <sub>t</sub>
10	36.43 <sub>s</sub>	35.65 <sub>s</sub>	20	16.21 <sub>q</sub>	16.26 <sub>q</sub>

4 $\alpha$ -Me), 1.23 (s, 3H, 13 $\beta$ -Me), 3.42 (dd, 1H,  $J$  = 8 Hz and 4 Hz, H-3), 3.37 and 4.23 (doublets,  $J$  = 10.75 Hz, 4 $\beta$ -CH<sub>2</sub>OH), 4.81 and 4.86 (six lines pattern, 2H,  $J$  = 10.5 Hz, 17.5 Hz, 1.5 Hz, -CH=CH<sub>2</sub>), 5.71 (dd, 1H,  $J$  = 10.5 Hz and 17.5 Hz, -CH=CH<sub>2</sub>)  
<sup>13</sup>C NMR (CDCl<sub>3</sub>, 75.45 MHz) see Table 1

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#### REFERENCES

1. Passannanti, S., Paternostro, M. P., Piozzi, F. and Barbagallo, C. (1984) *J. Nat. Prod. (Lloydia)* **47**, 885.
2. Brullo, S. and Furnari, F. (1979) *Webbia* **34**, 439.
3. Polonsky, J., Baskevitch, Z., Cagnoli-Bellavita, N. and Ceccherelli, P. (1970) *Bull. Soc. Chim. Fr.* 1912.

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## A DITERPENE WITH A NEW CARBON SKELETON FROM *SOLIDAGO ALTISSIMA*

Ferdinand Bohlmann, Pahup Singh\*, Ravindra K. Singh\*, Krishna C. Joshi\* and Jasmin Jakupovic

Institute for Organic Chemistry, Technical University of Berlin, D-1000 Berlin 12, West Germany, \*Department of Chemistry, University of Rajasthan, Jaipur, India

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**Key Word Index**—*Solidago altissima*, Compositae, diterpenes, new carbon skeleton

**Abstract**—An Indian sample of *Solidago altissima* afforded in addition to several clerodanes already isolated from other *Solidago* species, a diterpene with a new carbon skeleton. Furthermore a ketone and a new anethole derivative were present.

*Solidago altissima* has been studied previously. In addition to widespread compounds, several clerodanes were isolated [1]. We now have studied a sample collected at Jaipur, India. The aerial parts gave 1 $\beta$ -hydroxygermacra-4(15),5,10(14)-triene [2], the known *cis*-clerodane lactones solidago lactone II [1], III [1], V [1] and VII [1] as well as the corresponding tiglate [3].

The roots also contain solidago lactone II, III, V, VII and the corresponding tiglate as well as the precursors of the latter two, where the epoxide was replaced by a 3,4-double bond [5] and 2-oxo-kolavenic acid lactone, already prepared from the corresponding alcohol [4]. Furthermore the angelate 2, the *trans*-clerodanes kolavenic acid [5], 7 $\alpha$ -acetoxykolavenic acid [6], and 2-oxo-kolavenic acid [7] were present together with the aldehyde 1. The structure of the anethole derivative 2 clearly followed from the spectral data (see Experimental).

The acid 1 was transformed to the methyl ester 1a. The molecular formula was C<sub>21</sub>H<sub>32</sub>O<sub>3</sub> and the <sup>1</sup>H NMR spectral data (see Experimental) indicated the presence of an aldehyde by the singlet at  $\delta$  9.96. The similarity of several signals with those of methyl kolavenoate showed that most likely these two diterpenes were closely related. The presence of a homoallylic coupling between H-18 and H-1 required a five membered ring, especially as by careful

spin decoupling all signals and, accordingly, all sequences could be established. Therefore kolavenic acid is the precursor of 1, which may be formed via oxidative cleavage followed by aldol condensation. Alternatively a rearrangement of 2-hydroxy-3,4-epoxy kolavenic acid can be considered. Compound 1 we have named solidagonal acid.

The co-occurrence of *cis*- and *trans*-clerodanes in this species is remarkable. In all cases, where a 6-oxygen function is present, the configuration is 9 $\beta$ -methyl and 10 $\alpha$ H, while in all other cases *trans*-clerodanes with 10 $\beta$ H and 9 $\alpha$  methyl were observed.

#### EXPERIMENTAL

The air dried aerial parts (200 g) and the roots (70 g), collected in the campus of Rajasthan University, Jaipur, India, voucher deposited in RUBL-Herbarium, were extracted with Et<sub>2</sub>O-petrol, 1:2, and worked-up in the usual fashion [8]. CC fractions of the aerial parts (A) and the roots (R) were as follows: 1 (petrol), 2 (Et<sub>2</sub>O-petrol, 1:4), 3 (Et<sub>2</sub>O-petrol, 1:1) and 4 (Et<sub>2</sub>O). Fraction 2A (aerial parts) gave 15 mg 1 $\beta$ -hydroxygermacra-4(15),5,10(14)-triene. TLC of 3A (CH<sub>2</sub>Cl<sub>2</sub>-C<sub>6</sub>H<sub>6</sub>-Et<sub>2</sub>O, 4:4:1) gave 40 mg solidago lactone VII (R<sub>f</sub> 0.5), 50 mg of the corresponding tiglate (R<sub>f</sub> 0.4) and 10 mg solidago lactone V (R<sub>f</sub> 0.7).